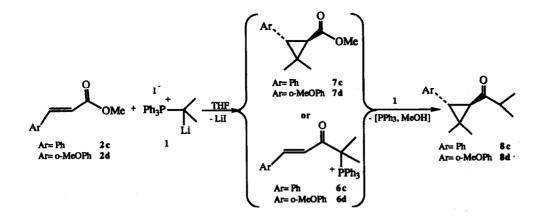
On the Reactivity of i-Propylidenetriphenylphosphorane with some α , β -Unsaturated Esters, -Amides and Acyl Chlorides

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Abstract: Isopropylidenetriphenylphosphorane allows the cyclopropanation of α , β -unsaturated carbonyl compounds including α , β -unsaturated esters, -ketones, -amides and -acid chlorides. Concomitant reaction on the carbonyl group of methyl cinnamates and cinnamoyl chlorides leading to cyclopropyl ketones has been observed.

In the course of a work directed towards the synthesis of chrysanthemic esters, 1-7 we had the occasion to react 1 isopropylidenetriphenylphosphorane 1, generated from the corresponding phosphonium iodide and n-BuLi, with methyl crotonate 2a, -hexenoate 2b and -cinnamate 2c used as models. Whereas the former derivative did not produce a definite product, probably due to a competing enolisation reaction, the second led chemoselectively to the corresponding methyl gem-dimethyl cyclopropane carboxylate 7b in moderate yield and the later surprisingly delivered a mixture of the expected methyl cyclopropane carboxylate 7c beside substantial amounts of (2,2-dimethyl-3-phenyl)cyclopropyl isopropyl ketone & (Schemes 1, 2 entries 3, 4). This type of reactivity seems to be general for those α,β -unsaturated methyl esters bearing an aryl group at the β -site since we observed similar results from the o-methoxy phenyl substituted cinnamate 2d (Scheme 2 entries 5)



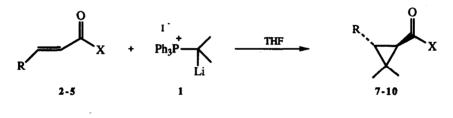


These ketones might arise from the reaction of isopropylidenetriphenylphosphorane 1 (i) on the carbonyl group of methyl cinnamates 2c, 2d prior to their cyclopropanation or (ii) on the carbonyl group of methyl 2-aryl-

3,3-dimethylcylopropane carboxylates 7c, 7d just after the cyclopropanation has taken place. We have ruled out the second alternative since we found that isopropylidenetriphenylphosphorane 1 does not react further with methyl 2-phenyl-3,3-dimethyl carboxylate 7c even under forced conditions.

The ester 7c required for this control experiment cannot be obtained directly from the corresponding methyl cinnamate 2c and the phosphonium ylide 1 since we have been unable to separate it, by distillation or chromatography on SiO₂, from the cyclopropyl ketone & concomitantly produced. We have however found that this ester can be chemoselectively produced, after trans-esterification, from the same ylide and isopropyl cinnamate 2e whose carbonyl group is more hindered than the previous one's (Scheme 2, entry 6).

These results led us to investigate the reaction of isopropylidenetriphenylphosphorane 1 with 4-methyl-1phenyl-penten-3-one 3c (Scheme 2, entry 7),⁸ α , β -unsaturated N,N-dimethyl amides 4 including the cinnamoyl amide 4a (Scheme 2, entries 8-13), and cinnamoyl chlorides 5c and 5f (Scheme 2, entries 14, 15), whose electrophilicity as well as steric hindrance around the carbonyl group differ from those of the esters described above.



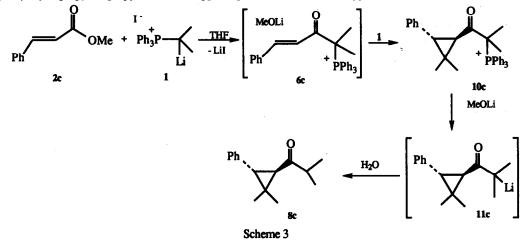
Scheme 2

Entry	x	R	Starting material	Experimental conditions (equiv.)	Yield % in 7-10 (X)
1	OMe	Me	2a	1 (1.1), THF, -78°C, 1h, 20°C, 1.5h	traces, 7a (OMe)
2	OMe	Pr	2 b	1 (1.1), THF, -78°C, 5h, 20°C, 3.5h	46%, 7b (OMe)
3	OMe	Ph	2 c	1 (1.1), THF, 0°C, 1h, 20°C, 3.5h	31%, 7c (OMe); 7%, 8c (i-Pr)
4	OMe	Ph	2 c	1 (1.3), THF, 0°C, 1h, 20°C, 5h	45%, 7c (OMe); 21%, 8c (i-Pr)
5	OMe	o-MeO-Ph	2d	1 (1.3), THF, 0°C, 1h, 20°C, 12h	33%, 7d (OMe); 12%, 8d (i-Pr)
6	Oi-Pr	Ph	2 e	1 (1.3), THF, 0°C, 1h, 20°C, 2.4h	70%, 7e(O-i-Pr)
7	i-Pr	Ph	3c	1 (1.2), THF, 0°C, 1h, 20°C, 2h	67%, 8c (i-Pr)
8	NMe2	Me	4a	1 (1.1), THF, 0°C, 1h, 20°C, 4h	0%, 9a (NMe2)
9	NMe ₂	Pr	4 b	1 (1.2), THF, 20°C, 6.5h	20%, 9b (NMe ₂)
10	NMe ₂	Pr	4 b	1 (2), THF, 20°C, 1h, 70°C, 3h	60%, 9b (NMc ₂)
11	NMe ₂	Ph	4 c	1 (1.3), THF, 0°C, 1h, 20°C, 3.5h	38%, 9c (NMe2)
12	NMe ₂	Ph	4c	1 (1.3), THF, 0°C, 1h, 20°C, 48h	48%, 9c (NMe2)
13	NMe2	Ph	4 c	1 (2.4), THF, 0°C, 1h, 20°C, 4h	67%, 9c (NMe2)
14	Cl	Ph	5 c	1 (2.4), THF, 0°C, 1h, 20°C, 4h	30%, 8c (i-Pr); 63%, 10c (C(Me)2+PPh3), I -
15	Cl	p-NO2-Ph	5f	1 (2.4), THF, 0°C, 1h, 20°C, 4h	46%, 8f (i-Pr); 46%, 10f (C(Me)2+PPh3), I -

The ketone 3c reacted efficiently (Scheme 2, entry 7) whereas the amides 4 proved, as expected, much less reactive and required quite drastic conditions for successful cyclopropanation (Scheme 2, entries 9-13). Both

reactions were completely chemoselective leading to an analytically pure authentic sample of (2,2-dimethyl-3phenyl)cyclopropyl isopropyl ketone & (Scheme 2, entries 7) and to the cyclopropane carboxamides 9b and 9c (Scheme 2, entries 10;13) in reasonably good yields. Again the crotonic acid derivative 4a did not deliver the corresponding cyclopropane derivative (Scheme 2, entry 8 compare to entry 1).

The case of cinnamoyl chlorides 5c and 5f (Scheme 2, entries 14 and 15) required further comments since these compounds do not lead to 2,2-dimethyl-3-aryl cyclopropane carboxylic acid chlorides nor to the corresponding acids (by hydrolysis) but provided instead a readily separable mixture of the (2,2-dimethyl-3phenyl)cyclopropyl isopropyl ketones 8c and 8f and of (2,2-dimethyl-3-phenyl)cyclopropyl 2-(2triphenylphosphino)propyl ketones 10c and 10f which are expected to be their direct precursors. Treatment of the phosphonium salt 10c with lithium methylate (1.5 equiv., MeOH, 20°C, 48h), the by-product expected to be formed concomitantly to (2,2-dimethyl-3-phenyl)cyclopropyl 2-(2-triphenylphosphino)propyl ketone 10c on reaction of isopropylidenetriphenylphosphorane with methyl cinnamate, led to the formation of (2,2-dimethyl-3phenyl)cyclopropyl isopropyl ketone 8c supporting thus the mechanism suggested in the Scheme 3.



To our knowledge they are only few reports dealing with the reaction of phosphonium ylides with acid chlorides or carboxamides and even fewer involve their α , β -unsaturated analogues.⁹ The reaction we have reported on methyl cinnamate does not seem to be precedented.^{2-7,9-13} In fact if reactions of phosphonium ylides on the carbonyl group of carboxylic esters or acid chlorides have been from time to time reported,⁹ they all deal with ylides bearing at least one hydrogen on the carbanionic center and the resulting β -keto phosphonium salts, which possess particularly acidic hydrogens suffer from a metallation reaction rather than from the reaction reported here. Work is now in progress in order to use this novel reaction for the regioselective synthesis of highly substituted enolates.

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